Asymmetric Catalytic Cyclopropanation of Olefins: Bis-Oxazoline Copper Complexes

Richard E. Lowenthal, Atsushi Abiko, and Satoru Masamune* Department of Chemistry, Massachusetts Institute of Technology Cambridge, Massachusetts 02139

Summary: Bis-oxazolines (1-7) prepared from diethyl malonate and chiral amino alcohols were converted into their Cu(II) complexes which, upon activation, exhibit high enantioselectivity of up to 99 %ee for catalytic cyclopropanation of olefins.

Over more than two decades copper-catalysed cyclopropanation of olefins with a diazoacetate has received much attention of synthetic chemists,¹ and a variety of chiral ligands for the metal have been devised for its asymmetric version as represented by (1) iminodiols prepared from salicylaldehydes with chiral amino alcohols (Aratani),² (2) 3-trifluoroacetyl-(+)- or (-)-camphor (Matlin),³ and (3) chiral semicorrin derivatives (Pfaltz).⁴ We wish to add to this list⁵ the chiral bis-oxazoline Cu(II) complex **1a** (see below) which, upon activation, catalyzes cyclopropanation with enantioselectivities that, at least equal or often exceed those achieved by the known catalysts. Ligand **1** which obviously is patterned after the semicorrin, is amenable to practical synthesis and also can be readily and extensively modified for other asymmetric reactions which are currently under investigation in our laboratories.

Several bis-oxazoline ligands (1-7) were prepared in one step from diethyl malonate and the corresponding amino alcohols and were converted into their violet Cu(II) complexes 1a-7a (see below for details). Treatment of the complexes with phenylhydrazine provided the active catalytic species which were evaluated in terms of stereoselectivity for cyclopropanation using ethyl diazoacetate and styrene in a standard fashion.^{2d} The results are summarized in Table 1. Expectedly its enantioselectivities of both trans- and cis- products increase as the size of the R group in the ligands is enlarged and both 90 %ee for the trans and 77 %ee for the cis with 1a compare favorably with those



achieved with a semicorrin (85 %ee for trans, 68 %ee for cis) and an iminodiol (69 %ee for trans, 54 %ee for cis).



Table 1: Cyclopropanation of styrene with ethyl diazoacetate using 1a-7a.

<u>Ligand</u>		Diastereoselectivity	Enantioselectivity (%ee)			
	<u>Yield</u> a	<u>Trans : Cis</u> b	<u>Trans</u> C	Cis ^C	<u>Configuration</u> d	
2a	78%	71:29	28	30	(1S,2S)/(1R,2S)	
3a	88%	75:25	48	36	(1S,2S)/(1R,2S)	
4a	78%	72:28	19	31	(1S,2S)/(1R,2S)	
5a	72%	71:29	46	31	(1 R,2R)/(1S,2R)	
6a	76%	71:29	36	15	(1R,2R)/(1S,2R)	
7a	81%	70:30	60	52	(1R,2R)/(1S,2R)	
la	80%	75:25	90	77	(1R,2R)/(1S,2R)	

a) Isolated yield. b) Determined by GC. c) Determined by GC of the d(+)- and l(-)- menthyl esters. d) Assigned on the basis of the known optical rotation of the acids.

It is known that the alcohol moiety of diazoacetates also affects the stereoselectivity of cyclopropanation and this applies to the case of 1a as well. Thus, the trans:cis diastereoselctivity 75:25 with ethyl diazoacetate is enhanced to 86:14 with (-)-menthyl diazoacetate [and $(+)_{436}$ -1a] and more significantly the enantioselectivities for both diastereomers became nearly perfect in the latter case (Table 2). Finally, several typical olefins were selected to define the scope of asymmetric cyclopropanation with the 1a - menthyl diazoacetate pair. Table 3 summarizes the results which demonstate a consistantly high degree of stereoselectivity observed with this pair and further suggest a way 1a should be modified, if asymmetric cyclopropanation of certain olefin types needs to be further improved.

Table 2: Cyclopropanation with several diazoacetates using 1a.

<u>Diazo ester</u>	<u>Yield</u>	Diastereoselectivity <u>Trans : Cis</u>	Enantioselec <u>Trans</u>	tivity (%ee) <u>Cis</u>
Ethvl	80%	75:25	90	77
t-Butyl	73%	80:20	94	89
d(+)-Menthyl	71%	84:16	98	80
l(-)-Menthyl	72%	86:14	98	96

			Diastereoselectivity	Enantioselectivity (%ee)	
<u>Olefin</u>	Diazo ester	<u>Yield</u> a	Trans : Cis ^b	<u>Trans</u> ^c	Cisc
styrene	l(-)-menthyl	72%	86:14	98	96
-	d(+)-menthyl	71%	84:16	98	80
1-octene	l(-)-menthyl	76%	94:6	99*	30*
	d(+)-menthyl	72%	90:10	75*	45*
a-methylstyrene	l(-)-menthyl	78%	89:11	92	79*
	d(+)-menthyl	72%	85:15	83	77*
trans-4-octene	l(-)-menthyl	52%		88	
	d(+)-menthyl	50%		85	
trans-anethole	l(-)-menthyl	45%	98:2	90	
	d(+)-menthyl	48%	95:5	88	65
2,3,3-trimethylbutene	l(-)-menthyl	60%	95:5	80	91
•	d(+)-menthyl	55%	98:2	77	
	•			,	

Table 3: Cyclopropanation using 1a and a variety of olefins.

a) Isolated yield. b) Determined by GC. c) Determined by GC of the d(+)- and l(-)- menthyl esters.

*) Enantiomeric excess was determined by conversion to the (R)-(-)-2-octyl ester .

<u>Preparation of 1.</u> To a solution of 4.75 g (40.6 mmol) *t*-leucinol in 162 mL of dry xylene was added 3.08 mL (20.3 mmol) of diethyl malonate. The mixture was heated to reflux for 4 h with a Dean-Stark trap. To this solution was added 89 mg (0.4 mmol) of dichlorodimethlystannane⁶ and the mixture was heated to reflux for an additional 48 h. The reaction mixture was concentrated to ca. 15 mL and chromatographed on 150 g of neutral alumina (Aldrich grade I) using ethyl acetate/hexane (3:2). The combined fractions were concentrated and the residue was bulb-to-bulb transferred to provide 4.75 g (88%) of crystalline bis-oxazoline 1, mp 48-49° C, $[a]_D$ -120° (c 0.51, CHCl₃). Further crystallization from CH₂Cl₂/Hexane did not change physical data.

Preparation of 1a. Deprotonation of 532 mg (2mmol) of 1 in 8 mL of THF was effected with 0.8 mL (2 mmol) of 2.5 M n-BuLi for 30 min at -78° C and then for 5 min at 25° C. To this solution was added 134.5 mg (1 mmol) of CuCl₂ and the mixture was stirred for 4 h. The solution was concentrated to 1 mL and the residue was chromatographed on 75 g of Aldrich neutral alumina (grade I) to provide 572 mg (96%) of dark purple crystals. Recrystallization from cyclohexane provided 512 mg (86%) of pure copper complex 1a which decomposed upon standing, mp 225° C (dec.), [a]_{436nm} +4978.6°. Anal. Calcd. C 60.63, N 9.43, H 8.48; Found C 60.40, N 9.50, H 8.37. mass spectra (EI) M⁺ m/z 593 (47.5%) and 595 (23.1%) for 63 Cu and 65 Cu.

<u>Typical cyclopropanation procedure.</u> Freshly prepared complex 1a (0.01 mmol) was combined with styrene (3 mmol) in dichloroethane (0.5 mL) at 25° C and was activated with phenylhydrazine (0.05 mL of a 1% solution).^{2d} After 5 min ethyl diazoacetate (1 mmol) in dichloroethane (1 mL) was added over a 24 h period. The reaction was concentrated and the residue was purified by chromatography. Enantiomeric excess was determined by gas chromatography of the menthyl ester, as described by Pfaltz, et. al.⁴

Acknowledgement. We are indebted to Mr. John C. Roberts and Ms. Sandra A. Filla for constructive suggestions and technical assistance and to the National Institutes of Health, U.S.A. (GM35879) and Kao Corporation, Japan for financial support.

References

1. Maas, G. "Topics in Current Chemistry", Springer-Verlag, Vol. 137, 1987, pp75-253.

 (a) Aratani, T.; Yoneyoshi, Y.; Nagase, T. Tetrahedron Lett. 1975, 1707. (b) Aratani, T; Yoneyoshi, Y.; Nagase, T. Tetrahedron Lett. 1977, 2599. (c) Aratani, T.; Yoneyoshi, Y.; Nagase, T. Tetrahedron Lett. 1982, 23, 685. (d) Aratani, T. Pure and Appl. Chem. 1985, 57, 1839.

3. Matlin, S. A.; Lough, W. J.; Chan, L; Abram, D. M. H.; Zhou, Z. J.C.S. Chem. Commun. 1984, 1038.

(a) Fritschi, H.; Leutenegger, U.; Siegmann, K.; Pfaltz, A. Helv. Chim. Acta 1988, 71, 1541. (b)
Fritschi, H.; Leutenegger, U.; Pfaltz, A.; Keller, W.; Kratky, C. Helv. Chim. Acta 1988, 71, 1553. For a recent review, see; Pfaltz, A. Mod. Synthetic Methods 1989, 198-248 (R. Scheffold, Ed., Springer Berlin).
In addition to references 2-4, see (a) Nozaki, H.; Takaya, H.; Moriuti, S.; Noyori, R. Tetrahedron 1968, 24, 3655. (b) Krieger, P.E.; Landgrebe, J.A. J. Org. Chem. 1978, 43, 4447. (c) Holland, D.; Laidler, D. A.; Milner, D. J. J. Mole. Catal. 1981, 11, 119. (d) Baldwin, J. E.; Carter, C. G. J. Am. Chem. Soc. 1982, 104, 1362. (e) Laidler D. A.; Milner, D. J. J. Organomet. Chem. 1984, 270, 121. (f) Brunner, H.; Goldbrunner, J. Chem. Ber. 1989, 122, 2005. For the use of Co(II) complexes, see (g) Tatsuno, Y.; Konishi, A.; Nakamura, A.; Otsuka, S. J.C.S. Chem Commun. 1974, 588. (h) Nakamura, A. Pure and Appl. Chem. 1978, 50, 37. (i) Nakamura, A.; Konishi, A.; Tatsuno, Y.; Otsuka, S. J. Am. Chem. Soc. 1978, 100, 3443. (j) Nakamura, A.; Konishi, A.; Tsujitani, R.; Kudo, M.; Otsuka, S. J. Am. Chem. Soc. 1978, 100, 3449.

6. Stetin, C.; Jeso, B.; Pommier, J. C. Synthetic Comm. 1982, 12, 495.

(Received in USA 16 July 1990)